



Terry, L. R. (2020). *Confining Hydrogen: A Low Energy Route to Room Temperature Superconductivity*. Poster session presented at STEM for Britain 2020, London, United Kingdom.

Peer reviewed version

[Link to publication record in Explore Bristol Research](#)
PDF-document

University of Bristol - Explore Bristol Research

General rights

This document is made available in accordance with publisher policies. Please cite only the published version using the reference above. Full terms of use are available:
<http://www.bristol.ac.uk/red/research-policy/pure/user-guides/ebr-terms/>

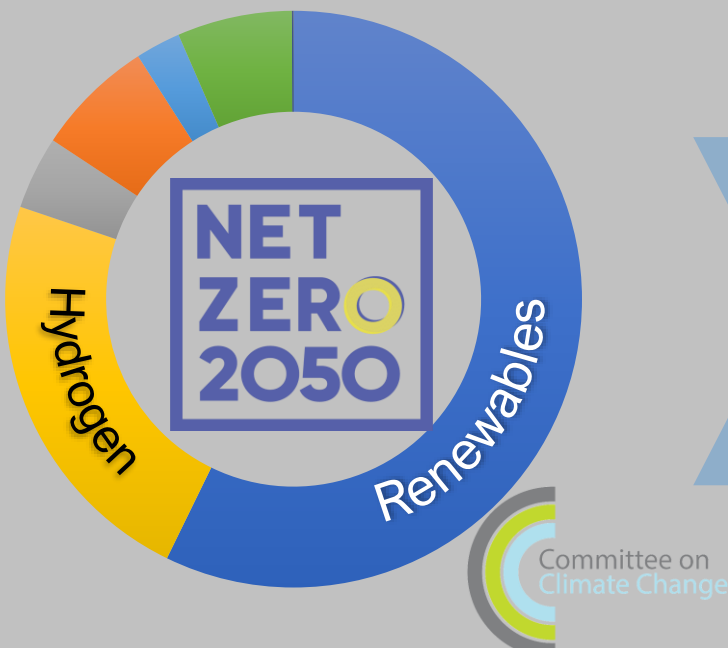
CONFINING HYDROGEN: A Low Energy Route to

Room Temperature Superconductivity

Lui Terry and Valeska Ting

Advanced Composites Collaboration for innovation and Science (ACCIS), University of Bristol, UK

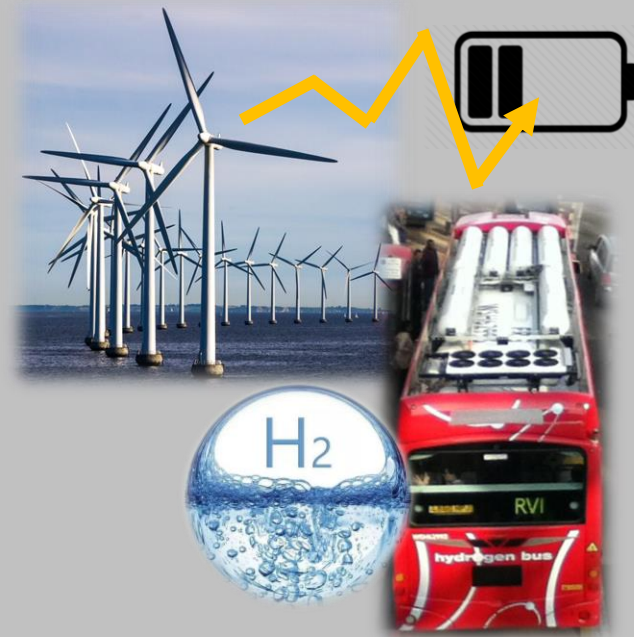
Context



Net Zero

- By 2050 UK emissions must reach net zero.
- **Renewable energy** sources need to reach around 57 % of our energy production.
- Transition to **hydrogen** as a non-polluting alternative to fossil fuels.

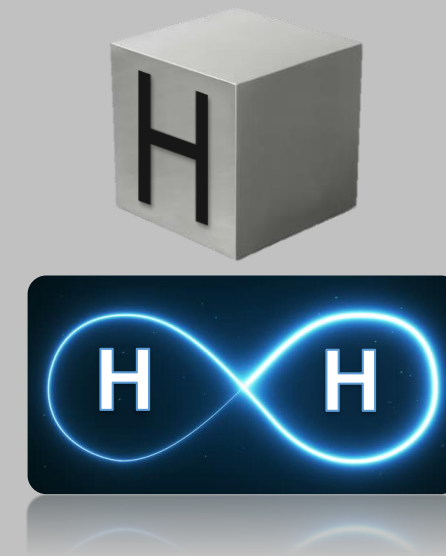
Problems



eStorage

- Intermittent renewable energy necessitates **long-term energy storage**.
- Chemical storage in hydrogen requires bulky pressurised gas tanks.
- Need compact and safe **energy storage**.

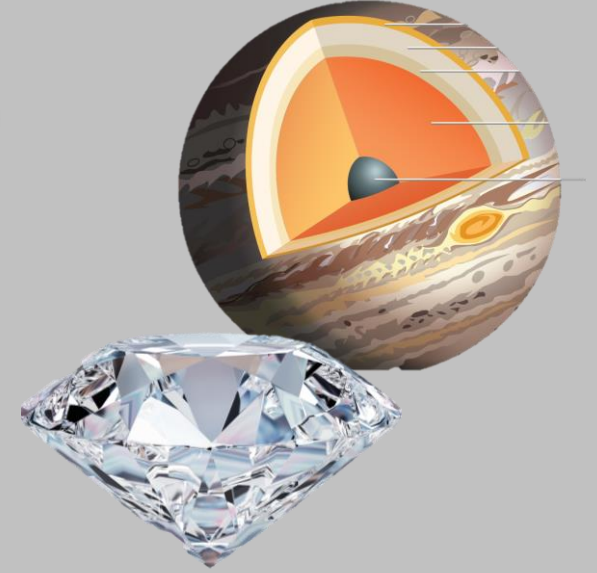
Solution



Metallic Hydrogen

- Densest form of solid hydrogen - a **supremely compact** way to store clean fuel.
- Conducts electricity with zero resistance at room temperature (a **superconductor!**).
- Renewable electricity could be stored in superconductive rings with **no power loss**.

Technical Challenge

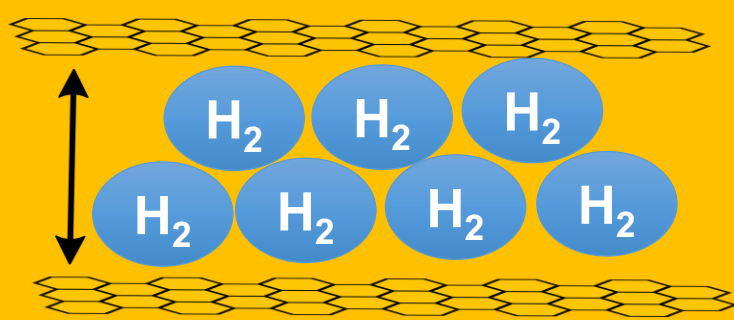


Pressure

- Metallic hydrogen is formed at **very high pressures** – close to the centre of a planet!
- Energy intense and currently impractical to produce...
- Need **an alternative way** to reach solid densities at lower pressures and higher temperatures.

Solution

DENSE HYDROGEN



Via **NANOCONFINEMENT**

Previous experimentation indicates that confining hydrogen in very small carbon pores produces solid-like densities at **pressures ~2000 times lower** than classically observed!

“Route to Room Temperature Superconductivity”

Using nanoconfinement as a ‘**pre-densification**’ step and the application of additional pressure, may allow us to **create metallic hydrogen** at significantly lower pressures than the centre of a planet.

Density of hydrogen could be increased by confining in porous carbon sponges



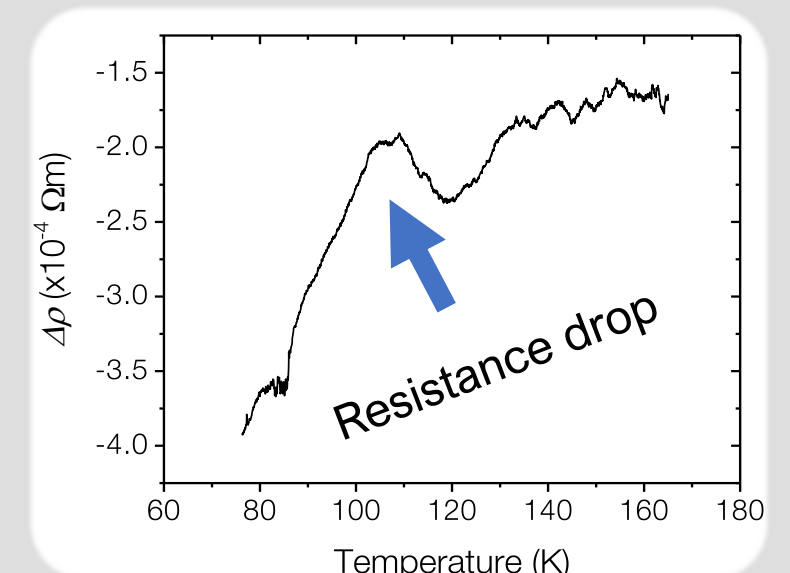
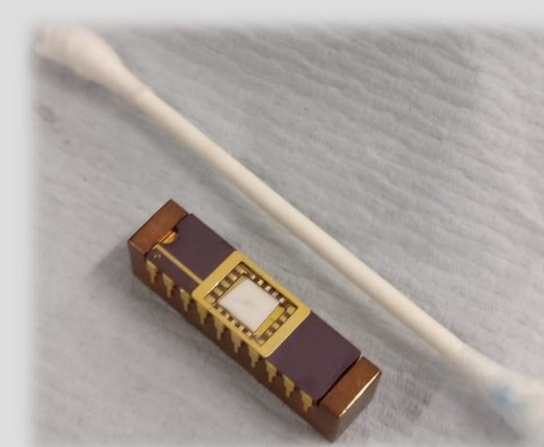
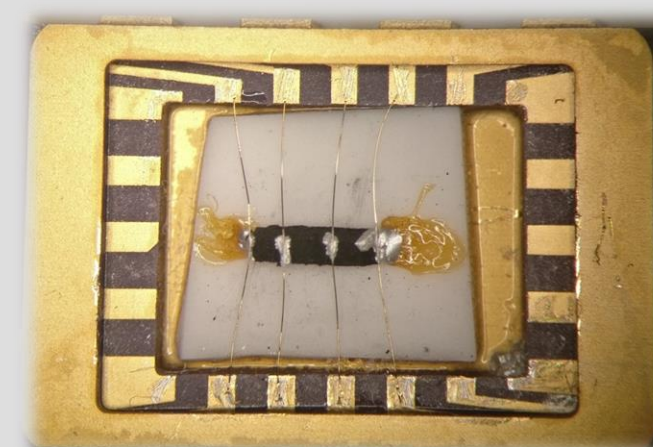
1) Proof of Principle – It is Solid!

- **Neutron Diffraction** reveals structural information of a material as well as definitive proof of its state.
- Our recent studies show for the first time, the crystal structures of nanoconfined hydrogen - fully **confirming that dense-solid hydrogen** does form in carbon sponges. Furthermore, we observe them at temperatures never seen previously.



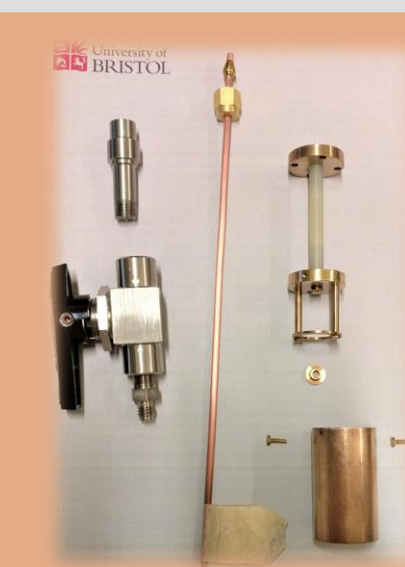
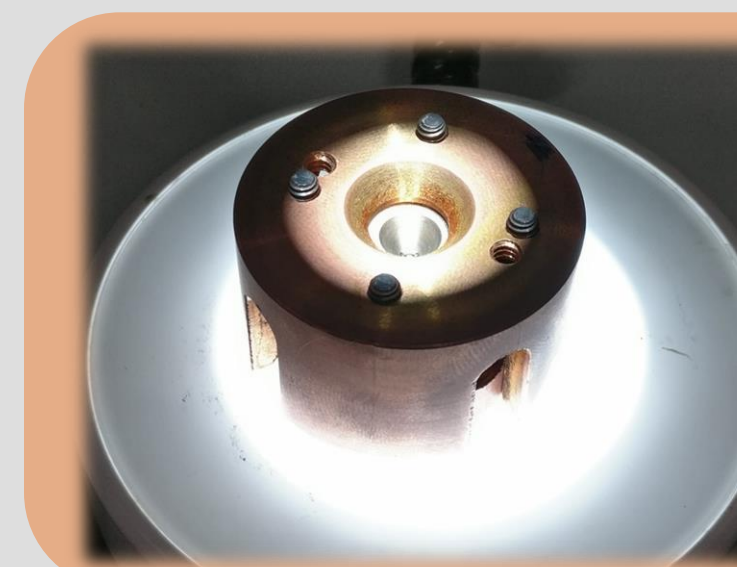
2) Zero Resistance

- **Electrical resistivity** measurements can identify a switch to superconductivity but also can identify changes in state.
- Low pressure results observed a **drop in resistivity** upon densification, signifying a switch in the hydrogen’s state.



What does this mean?

- Solid hydrogen = more hydrogen = **compact fuel storage**.
- Lower pressures and higher temperatures = **safer storage**.
- Results so far are promising for the next stage of the project: increasing the pressure to find a **low energy route to room temperature superconductivity...**



Next Steps:
Building Pressure Construction of high pressure experimental apparatus